

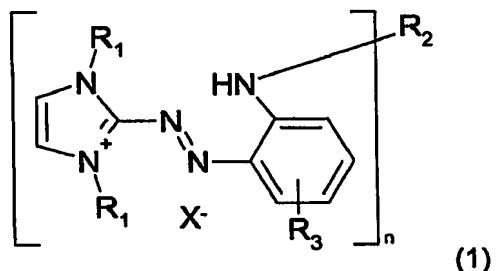
Cationic direct dyes

The present invention relates to novel cationic direct dyes, compositions thereof, to processes for their preparation and to their use in the dyeing of organic material, such as keratin fibers, wool, leather, silk, paper, cellulose or polyamides, especially keratin-containing fibers, cotton or nylon, and preferably hair, more preferably human hair.

It is known, for example, from WO 95/01772 that cationic dyes can be used to dye organic material, for example keratin, silk, cellulose or cellulose derivatives, and also synthetic fibers, for example polyamides. Cationic dyes exhibit very brilliant shades. A disadvantage is their unsatisfactory fastness to hydrolysis and to light, their frequently inadequate stability under reducing or oxidizing conditions, and their frequently unsatisfactory storage stability (see: John F. Corbett: "The Chemistry of Hair-Care Products", JSCD August 1976, page 290).

The actual technical problem of the present invention was to provide brilliant dyes that are distinguished by deep dyeing having good fastness properties with respect to washing, light, shampooing and rubbing, and that preferably exhibit satisfactory stability under reducing or oxidizing dyeing conditions, for the dyeing of organic material.

Accordingly, the cationic dye of formula (1)



has been discovered, wherein

R₁ is an unsubstituted or substituted C₁-C₁₄alkyl or an aryl radical;

X⁻ is an anion;

R₃ is an unsubstituted or substituted C₁-C₁₄alkyl, aryl radical, C₁-C₆alkoxy, cyanid, nitro or halide;

n is 1 or 2; and

if n is 1, then R₂ is hydrogen, unsubstituted or substituted C₁-C₁₄alkyl; or

if n is 2, then R₂ is unsubstituted or substituted C₁-C₁₄alkylen.

Preferred is a cationic dye of formula (1), wherein
R₁ is methyl.

In addition, more preferred is a cationic dye of formula (1), wherein
R₁ is methyl,
n is 1, and
R₂ is a substituted or unsubstituted C₁-C₁₂alkyl.

Further, more preferred is a cationic dye of formula (1), wherein
R₁ is methyl,
n is 2, and
R₂ is a substituted or unsubstituted C₁-C₁₄alkylen, preferred is a substituted or unsubstituted
C₁-C₈alkylen, and more preferred is a substituted or unsubstituted C₁-C₆alkylen.

Much more preferred is a cationic dye of formula (1), wherein
R₃ is hydrogen and
R₁ is methyl,
n is 1, and
R₂ is an unsubstituted or substituted C₁-C₁₂alkyl.

Further, much more preferred is a cationic dye of formula (1), wherein
R₃ is hydrogen and
R₁ is methyl,
n is 2, and
R₂ is a substituted or unsubstituted C₁-C₁₄alkylen, preferred is a substituted or unsubstituted
C₁-C₈alkylen, and more preferred is a substituted or unsubstituted C₁-C₆alkylen.

In the present invention, substituents of C₁-C₁₄alkyl or C₁-C₁₄alkylen, C₁-C₁₂alkyl or
C₁-C₁₂alkylen, C₁-C₈alkyl and C₁-C₈alkylen, and preferably C₁-C₈alkyl and C₁-C₈alkylen are,
for example hydroxyl; C₁-C₈alkyl, preferably methyl; or C₁-C₈alkoxy and C₁-C₈alkoxy; cyanide
and/or halide; or
NR₅R₆, wherein

R_6 and R_8 are each independently of the other hydrogen, unsubstituted or substituted aryl radical or C_1 - C_6 alkyl, preferably methyl, ethyl, propyl, isopropyl or t-butyl.

C_1 - C_{14} alkyl and C_1 - C_{14} alkylen, C_1 - C_{12} alkyl or C_1 - C_{12} alkylen, C_1 - C_8 alkyl and C_1 - C_8 alkylen, and preferably C_1 - C_6 alkyl and C_1 - C_6 alkylen; and may be straight-chain or branched, preferably branched alkyl or alkylen are tert.-butyl, isopropyl, dodecyl, 2,2-dimethylpropyl, or 2,2-dimethylpropylen; or preferably straight-chain alkyl or alkylen are ethyl, dodecyl, hexyl, ethylene, dodecylen, hexylen, octyl, octylen;

or

from C_5 alkyl or C_5 alkylen upwards, monocyclic or polycyclic, and preferably cyclohexyl or cyclohexylen;

and may be uninterrupted or interrupted by at least one hetero atom, such as O, S, N, NH; for example $-CH_2CH_2-O-CH_2CH_2-O-CH_2CH_3$, or $-CH_2CH_2-O-CH_2CH_3$, or $CH_2CH_2-O-CH_3$, or CH_2-O-CH_3 , or $CH_2CH_2CH_2-O-CH(CH_3)_2$, preferably $CH_2CH_2CH_2-O-CH(CH_3)_2$.

C_1 - C_{14} alkyl and C_1 - C_{14} alkylen, is C_1 - C_{12} alkyl or C_1 - C_{12} alkylen, C_1 - C_8 alkyl and C_1 - C_8 alkylen, and preferably C_1 - C_6 alkyl and C_1 - C_6 alkylen; and preferred is C_1 - C_{14} alkyl is C_1 - C_{12} alkyl, and C_1 - C_{14} alkylen is preferred C_1 - C_6 alkylen.

C_1 - C_{14} alkyl is, for example, methyl, ethyl, propyl, isopropyl, n-butyl, sec-butyl, tert-butyl, n-pentyl, 2-pentyl, 3-pentyl, 2,2'-dimethylpropyl, cyclopentyl, cyclohexyl, n-hexyl, n-octyl, 1,1',3,3'-tetramethylbutyl, 2-ethylhexyl, nonyl, decyl, undecyl or dodecyl.

C_1 - C_{12} alkyl is, for example, methyl, ethyl, propyl, isopropyl, n-butyl, sec-butyl, tert-butyl, n-pentyl, 2-pentyl, 3-pentyl, 2,2'-dimethylpropyl, cyclopentyl, cyclohexyl, n-hexyl, n-octyl, 1,1',3,3'-tetramethylbutyl, 2-ethylhexyl, nonyl, decyl, undecyl or dodecyl.

C_1 - C_8 alkyl is, for example, methyl, ethyl, propyl, isopropyl, n-butyl, sec-butyl, tert-butyl, n-pentyl, 2-pentyl, 3-pentyl, 2,2'-dimethylpropyl, cyclopentyl, cyclohexyl, n-hexyl, n-octyl, 1,1',3,3'-tetramethylbutyl or 2-ethylhexyl.

C_1 - C_6 alkyl is, for example, methyl, ethyl, propyl, isopropyl, n-butyl, sec-butyl, tert-butyl, n-pentyl, 2-pentyl, 3-pentyl, 2,2'-dimethylpropyl, cyclopentyl, cyclohexyl, n-hexyl.

C_1 - C_4 alkyl is, for example, methyl, ethyl, propyl, isopropyl, n-butyl, sec-butyl or tert-butyl.

Preferred is methyl, 2,2'-dimethylpropyl, n-hexylen

C_1 - C_{14} alkylen is, for example, methylen, ethylen, propylen, isopropylen, n-butylen, sec-butylen, tert-butylen, n-pentylen, 2-pentylen, 3-pentylen, 2,2'-dimethylpropylen, cyclopentylen, cyclohexylen, n-hexylen, n-octylen, 1,1',3,3'-tetramethylbutylen, 2-ethylhexylen, nonylen, decylen, undecylen or dodecylen.

C_1 - C_{12} alkylen is, for example, methylen, ethylen, propylen, isopropylen, n-butylen, sec-butylen, tert-butylen, n-pentylen, 2-pentylen, 3-pentylen, 2,2'-dimethylpropylen, cyclopentylen, cyclohexylen, n-hexylen, n-octylen, 1,1',3,3'-tetramethylbutylen, 2-ethylhexylen, nonylen, decylen, undecylen or dodecylen.

C_1 - C_8 alkylen is, for example, methylen, ethylen, propylen, isopropylen, n-butylen, sec-butylen, tert-butylen, n-pentylen, 2-pentylen, 3-pentylen, 2,2'-dimethylpropylen, cyclopentylen, cyclohexylen, n-hexylen, n-octylen, 1,1',3,3'-tetramethylbutylen or 2-ethylhexylen.

C_1 - C_6 alkylen is, for example, methylen, ethylen, propylen, isopropylen, n-butylen, sec-butylen, tert-butylen, n-pentylen, 2-pentylen, 3-pentylen, 2,2'-dimethylpropylen, cyclopentylen, cyclohexylen, n-hexylen.

C_1 - C_4 alkylen is, for example, methylen, ethylen, propylen, isopropylen, n-butylen, sec-butylen, tert-butylen,

C_1 - C_6 alkoxy is $O-C_1-C_6$ alkyl, preferably $O-C_1-C_4$ alkyl.

Aryl radical is, for example, an unsubstituted or substituted phenyl, benzyl- or toyl

Halide is, for example, fluoride, chloride, bromide or iodide, especially chloride and fluoride.

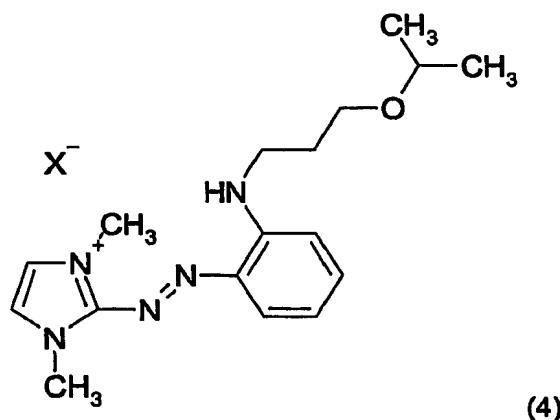
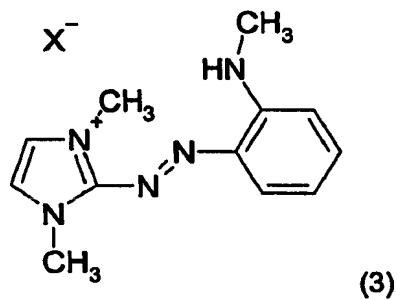
"Anion" denotes, for example, an organic or inorganic anion, such as halide, preferably chloride and fluoride, sulfate, hydrogen sulfate, phosphate, boron tetrafluoride, carbonate,

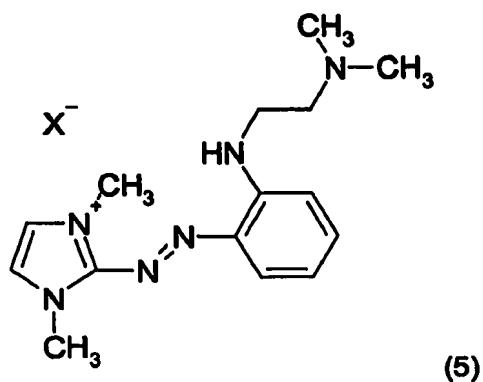
bicarbonate, oxalate or C₁-C₆alkyl sulfate, especially methyl sulfate or ethyl sulfate; anion also denotes lactate, formate, acetate, propionate or a complex anion, such as the zinc chloride double salt.

The anion is especially a halide, preferably chloride or fluoride, sulfate, hydrogen sulfate, methylsulfate, phosphate, formate, acetate or lactate.

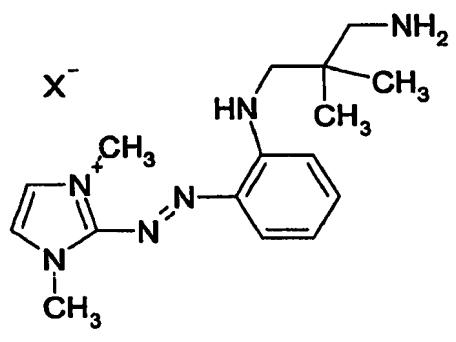
The anion is more especially chloride, methyl sulfate, formate or acetate.

More preference is given to cationic dyes of formulae (3), (4), (5), (6), (7), (8), (9), (10), (11), (12), (13), (14), (15), (16) or (17)

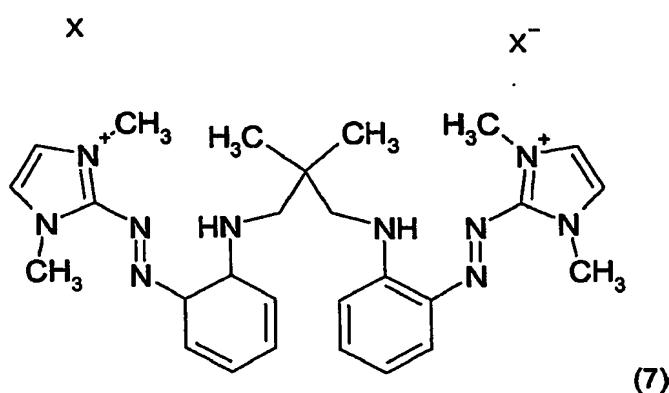




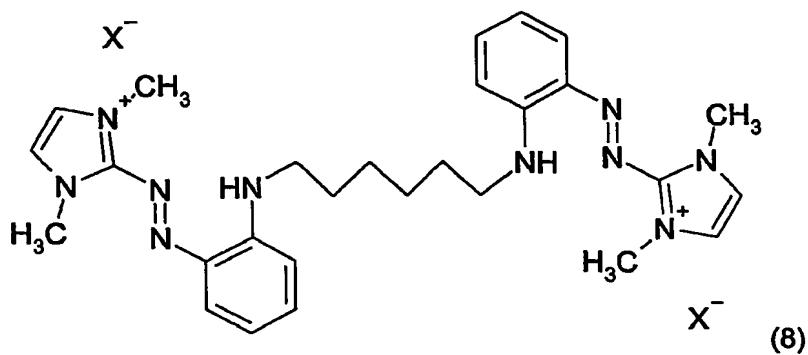
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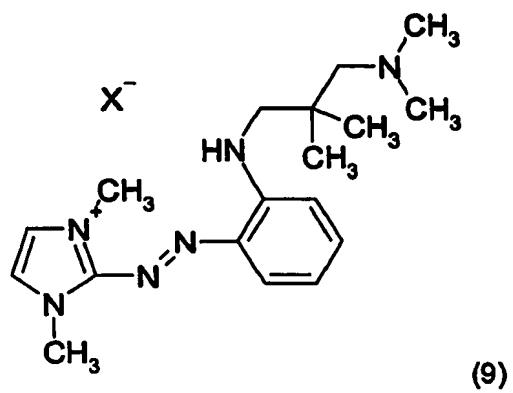
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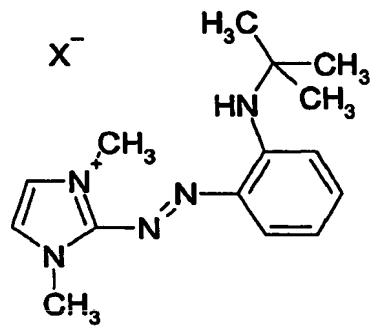
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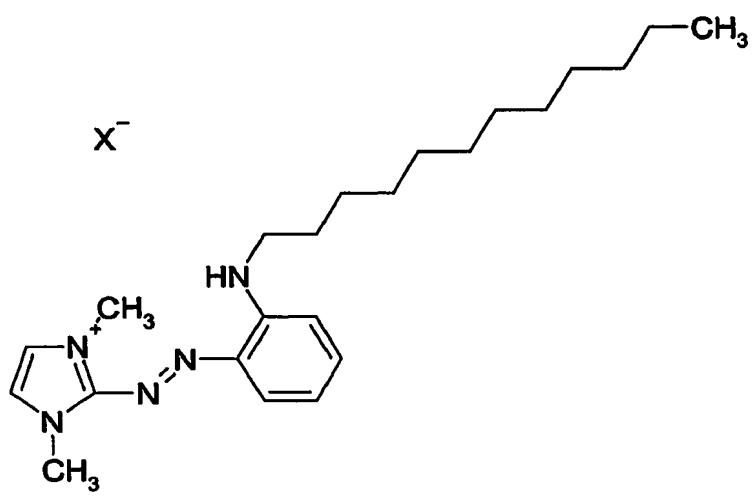
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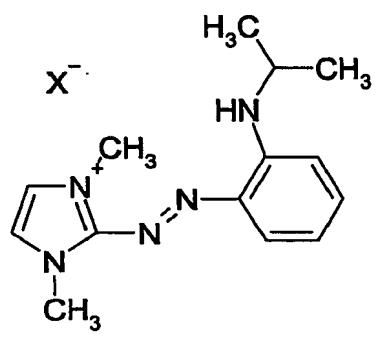
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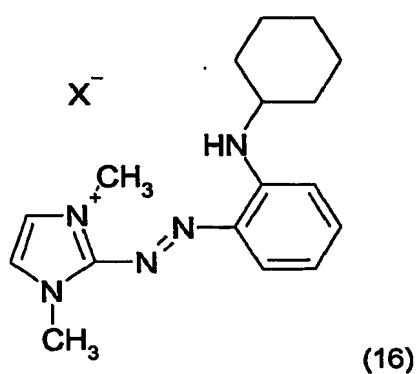
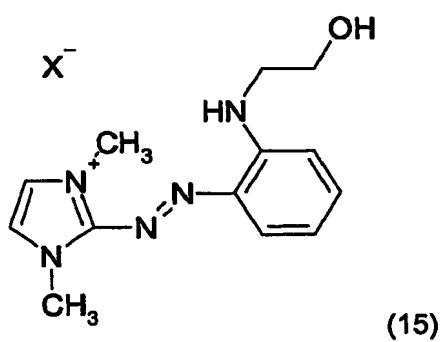
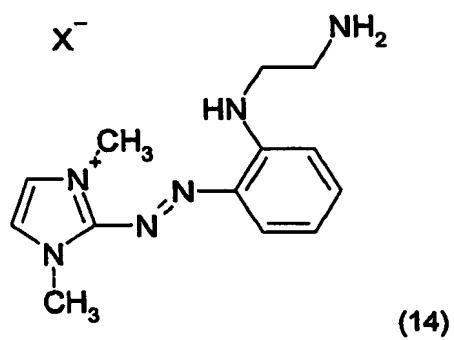
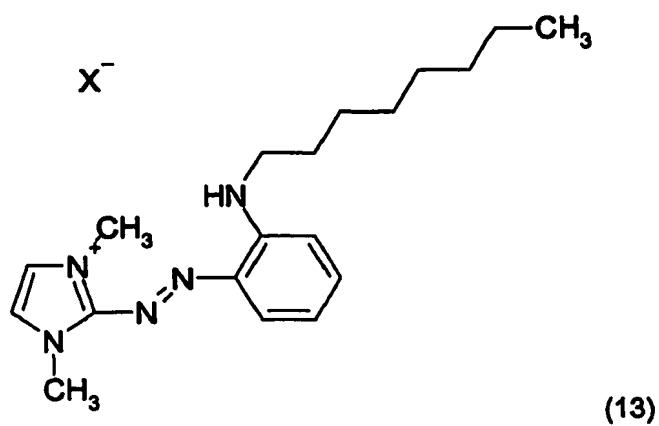
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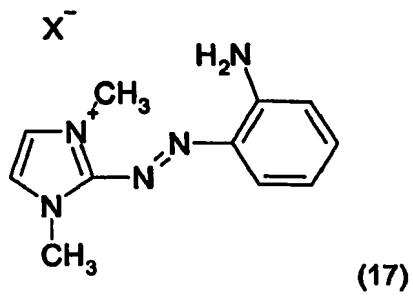


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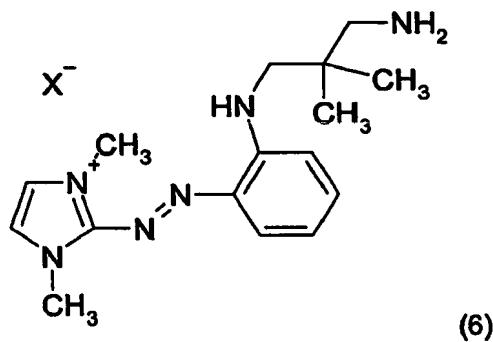




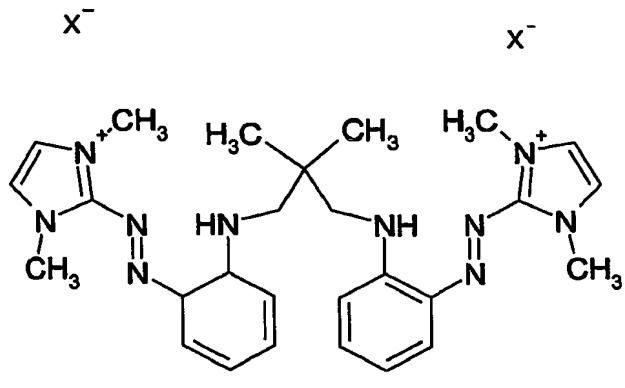
wherein

X^- is an anion.

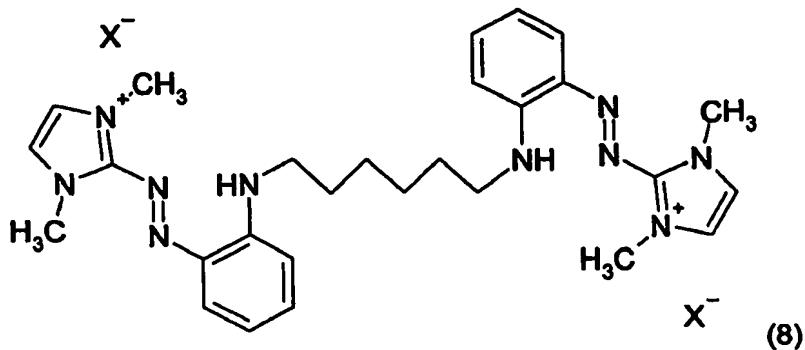
Most preference is given to cationic dyes of formulae (6), (7) or (8)



(6)



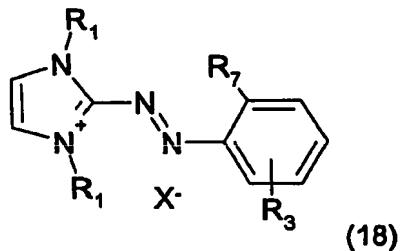
(7) , or



wherein

X⁻ is an anion.

The present invention concerns in addition a cationic dye of formula (18)



wherein

R₇ is C₁-C₆alkoxy or halide, preferred halides are chloride or fluoride, preferred fluoride, and R₁, R₃ and X⁻ having the same definition and preferences as given for compound of formula (1);

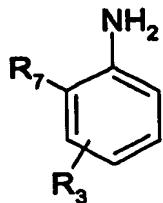
preferred is compound of formula (18), wherein R₇ is fluoride;

and more preferred is compound of formula (18), wherein R₇ is fluoride and R₁ is methyl;

and most preferred is compound of formula (18), wherein R₇ is fluoride, R₁ is methyl and R₃ is hydrogen.

Further, the present invention concerns a process for the preparation of cationic dyes of formula (18) as defined above, comprising

a) diazotiation of a compound of the following formula

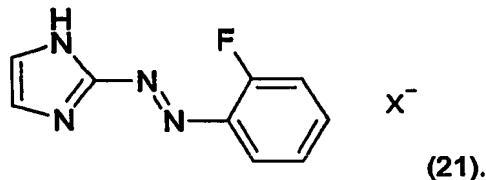


wherein

R₃ and **R₇** have the above given definitions and preferences, and

- coupling the diazotiated compound with imidazole, and
- then alkylation or arylation; preferably methylating.

The present invention also concerns a dye of formula (21)



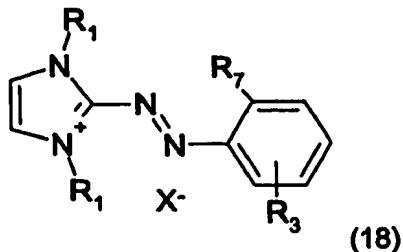
In addition, the present invention concerns a process for the preparation of cationic dyes of formula (21) as defined above, comprising

- diazotizing 2-fluoroaniline and
- then coupling with imidazole.

The above mentioned diazotation, coupling, alkylation and arylation are methods known per se, and are for example disclosed in standard textbooks, such as for example Jerry March, *Advanced Organic Chemistry: Reactions, Mechanismus and Structure*.

The present invention relates also to a process for the preparation of the dyes of formula (1), preferably to a dye of formula (3), (4), (5), (6), (7), (8), (9), (10), (11), (12), (13), (14), (15), (16) or (17) and comprises

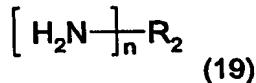
- bringing a compound of formula (18)



which is prepared according to the process described above,
and wherein

R7 is C1-C6alkoxy or halide, preferred halides are chloride or fluoride, preferred fluoride, and R1, R3 and X- having the same definition and preferences as given for compound of formula (1),

with an amine of formula (19)



wherein

n is 1 or 2; and

if n is 1, then R2 is hydrogen, unsubstituted or substituted C1-C14alkyl; or

if n is 2, then R2 is unsubstituted or substituted C1-C14alkylen;

into contact.

The reaction is generally initiated by bringing the compound of formula (18) and the amine of formula (19) into contact; for example by mixing together the starting compounds or by drop wise addition of one starting compound to the other.

Customary, the temperature is in the range of 290 to 300 K during the mixing of the starting compounds.

The molar ratio of compound of formula (18) to amine of formula (19) is generally selected in the range from 5:1 to 1:5, especially in the range from 3:1 to 1:3.

The duration of reaction is generally dependent on the reactivity of the starting compounds, on the reaction temperature chosen and on the desired conversion. The chosen duration of reaction is usually in the range from one hour to three days.

The reaction temperature for the reaction of the compound of formula (18) with the amine of formula (19) is advisable to select in the range from 293 to 363K, especially in the range from 323 to 355K, more especially in the range from 333 to 355K.

The reaction pressure chosen is generally in the range from 70 kPa to 10 MPa, especially from 90 kPa to 5 MPa, and is more especially atmospheric pressure.

It may be desirable to conduct the reaction of compound of formula (18) and amine of formula (19) in the presence of a base.

The molar ratio of compound of formula (18) to the base is generally selected in the range from 10:1 to 1:5, especially in the range from 1:1 to 1:3.

Suitable bases are for example an alkali metal hydroxide, such as sodium-, potassium or lithium hydroxide, such as sodium hydroxide, or an alkali metal carbonate, such as sodium-, potassium or lithium carbonate, such as sodium carbonate, or an alkali metal hydrocarbonate, such as sodium-, potassium or lithium hydrocarbonate, such as sodium-, potassium or lithium C₁-C₆alkyloxide, such as sodium-, potassium or lithium C₁-C₆alkyloxide, preferably sodium methoxide, potassium methoxide or lithium methoxide, or sodium ethoxide, potassium ethoxide or lithium ethoxide; or tertiary amines, for example, such as pyridine, trimethylamine, triethylamine, trioctylamine, 1,4-diazabicyclo[2.2.2]octan, chinuclidine, N-methylpiperidine; or alkalinmetal acetate, for example such as sodium acetate, potassium acetate, or lithium acetate.

Preferred are sodium hydroxide, sodium carbonate, tertiary amines, such as pyridine and 1,4-diazabicyclo[2.2.2]octan.

It may be desirable to conduct the reaction of compound of formula (18) and amine of formula (19) in the presence of a catalyst.

The molar ratio of compound of formula (18) to the catalyst is generally selected in the range from 100:0.01 to 100:10, especially in the range from 100:1 to 100:10, most preferred in the range of 100:4 to 100 to 6.

Suitable catalysts are for example an alkali metal C₁-C₆alkyloxide, such as sodium-, potassium or lithium C₁-C₆alkyloxide, preferably sodium methoxide, potassium methoxide or lithium methoxide, or sodium ethoxide, potassium ethoxide or lithium ethoxide; or tertiary amines, for example, such as pyridine, trimethylamine, triethylamine, trioctylamine, 1,4-diazabicyclo[2.2.2]octan, chinuclidine, N-methylpiperidine; or alkalinmetal acetate, for example such as sodium acetate, potassium acetate, or lithium acetate. Preferred are potassium acetate, sodium methoxide, pyridine and 1,4-diazabicyclo[2.2.2]octan.

It is recommended that the reaction mixture obtained be subsequently stirred in the selected temperature range from 293 to 363 K, especially in the range from 323 to 355K, more especially in the range from 333 to 355K. The duration chosen for the subsequent stirring is generally from 1 hour to 24 hours.

In addition, the reaction can be carried out with or without solvent, but is preferably carried out in a solvent. Preference is given to organic solvents or solvent mixtures.

Within the context of this invention, solvents are organic solvents and water, or a mixture of organic solvents or a mixture of organic solvents and water.

Organic solvents are, for example, protic or aprotic polar organic solvents, such as alcohols, for example methanol, ethanol, n-propanol, isopropanol, butanol or glycols, especially isopropanol, or nitrile, such as acetonitrile or propionitrile, or amide, such as dimethylformamide, dimethylacetamide or N-methylpyridine, or sulfoxide, such as dimethylsulfoxide, or mixtures thereof.

The ratio by weight of compound of formula (18) to the solvent is generally in the range from 20 to 90% by weight, especially in the range from 30 to 60% by weight.

The prepared product may be advantageously worked up and isolated, and if desired be purified.

Customary, the work up starts by decreasing the temperature of the reaction mixture in the range from 280 to 300 K, especially in the range from 290 to 300 K.

It may be advantageous to decrease the temperature slowly, over a period of several hours.

In general, the reaction product is usually filtered and then washed with water or a salt solution and subsequently dried.

Filtration is normally carried out in standard filtering equipment, for example Büchner funnels, filter presses, pressurised suction filters, preferably *in vacuo*.

The temperature for the drying is dependent on the pressure applied. Drying is usually carried out *in vacuo* at 50-200 mbar.

The drying is usually carried out at a temperature in the range from 313 to 363 K, especially from 323 to 353 K, and more especially in the range from 328 to 348 K.

It has proved advantageous for the product to be purified by recrystallisation after it has been isolated.

Organic solvents and solvent mixtures are suitable for the recrystallisation. Preference is given to alcohols, for example methanol, ethanol, 2-propanol or butanol, especially 2-propanol.

The dyes of formula (1) according to the invention are suitable for dyeing organic material, such as keratin, wool, leather, silk, paper, cellulose or polyamides, especially keratin-containing fibers, cotton or nylon, and preferably human hair.

The multiplicity of shades of the dye, which results by the method according to the present invention, can be increased by combination with other dyes.

The present invention relates also to the coloration of hair with a dye of formula (1) according to the present invention, and at least a single further dye.

The dye of formula (1) of the present invention can be combined with dyes of the same or different class of dyes, especially with direct dyes, oxidation dyes; dye precursor

combinations of a coupler compound and a diazotized compound, or a capped diazotized compound; and/or cationic reactive dyes.

Direct dyes are natural or synthetic; they are uncharged, cationic or anionic, such as acid dyes.

Oxidation dye denotes also for oxidation dye precursors, which are from the group of the developer and coupler compounds. Wherein the coupler compounds denotes also to the addition salts thereof with an acid.

In the context of the present invention the single classes of dyes comprise the dyes defined in the Color Index of the Society of Textile Chemist and Colorist.

Further, in the context of the present invention, combinations comprising of a compound of formula (1) are compositions, formulation, methods and.

The present invention further comprises compositions, formulation, methods or combinations comprising of a compound of formula (1).

Preferred for a combination with a cationic dye of formula (1) are cationic dyes, such as Basic Yellow 87, Basic Orange 31 or Basic Red 51, or cationic dyes as described in WO 01/66646, especially cationic dye of example 4, or cationic dyes as described in WO 02/31056, especially cationic dye of example 6, compound of formula 106 dye of formula (3) as described in EP-A-714,954.

In the present invention preferred does a composition comprise a compound of formula (1) and in addition at least a single further direct dye and/or an oxidative agent.

Further, in the present invention preferred does a composition comprise a compound of formula (1) and in addition at least a single oxidative dye and/or; at least a single oxidative dye and an oxidative agent.

In addition, in the present invention preferred is a composition comprising a compound of formula (1) in form of a shampoo, conditioner, gel or emulsion.

Further, in the present invention preferred is a method of dyeing organic material, especially human hair, that comprises bringing into contact with the organic material at least a single a cationic dye of formula (1).

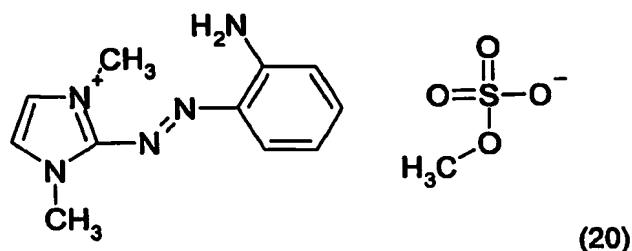
More preferred is in the present invention a method for dyeing or tinting human hair.

A further preferred embodiment of the present invention concerns a method for dyeing human hair or strands, that comprises contacting the hair with at least a single a cationic dye of formula (1) and an oxidative agent and, optionally, a further direct dye.

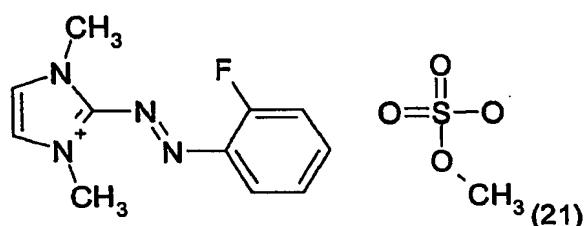
In addition, the present invention concerns a method for dyeing human hair according to any of claims 14 to 16, that comprises contacting the hair with at least a single a cationic dye of formula (1) and at least a single oxidative dye; or contacting the hair with a cationic dye of formula (1) and at least a single oxidative dye and an oxidative agent.

The dyes according to the invention are distinguished by brilliant shades. They are suitable for dyeing organic material, such as keratin, wool, leather, silk, paper, cellulose or polyamides, especially keratin-containing fibers, cotton or nylon, and preferably human hair. The dyeing obtained is distinguished by their depth of shade and their good fastness to washing properties, such as, for example, fastness to light, shampooing and rubbing. The stability and storage stability of the dyes according to the invention are excellent. They are accordingly especially suitable for dyeing under oxidizing and reducing conditions. The advantage of the new dyes according to the present invention, especially those of formula (4), (5), or (6), is their stability against reduction agents e. g. sodium sulfite and ascorbic acid. Therefore, they can be combined with oxidation dyes in one emulsion.

The following Examples serve to illustrate the processes for coloration without limiting the processes thereto. Unless specified otherwise, parts and percentages relate to weight. The amounts of dye specified are relative to the material being colored.

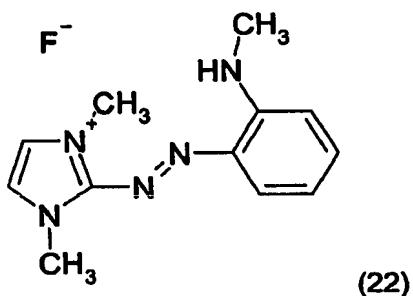
Examples A /Process of PreparationExample A1

12.4 g 2-Fluoroanillin is added to a stirred solution of 25 ml water and 25 ml of 32% hydrochloric acid at 295 K. Then, the reaction mixture is cooled to 273 K and 19 ml 36% sodium nitrite solution is dropped at such a rate that the temperature of the mixture is maintained in the range of 273 to 276 K. After the addition of the sodium nitrite solution, the mixture is stirred for one hour. If no excess of nitrite is detected during one hour (detection by using a potassium iodide/starch paper), further amounts of sodium nitrite solution is added. After this one hour the remaining excess of nitrite is destroyed with sulfamic acid. Then, the obtained diazo solution is dropped to a 273 K cold solution of 7.4 g imidazole in 30 ml water, whereby the pH of the solution is maintained in the range of pH 10 to 11 by adding 36% sodium hydroxide solution. After completing the diazo addition, the obtained suspension is warmed up to 295 K, the pH is adjusted to 10.5 with 36% sodium hydroxide solution. After one hour stirring at this pH and temperature, the suspension is filtrated. Then, the obtained filtercake is washed twice with 50 ml water to obtain 55 g of the humid product. Then, the humid product is suspended in 500 ml water and 0.3 mole dimethylsulphate and sodium hydroxide added simultaneously for maintaining the pH at 10-10.3 and the temperature at 25 to 30°C. The mixture is hold for one hour, to finish the hydrolysis of excess of dimethylsulphate. The the water is evaporated and about 40 g humid solid obtained, which gives 27 g dried (0.07 mole) product with the following formula:



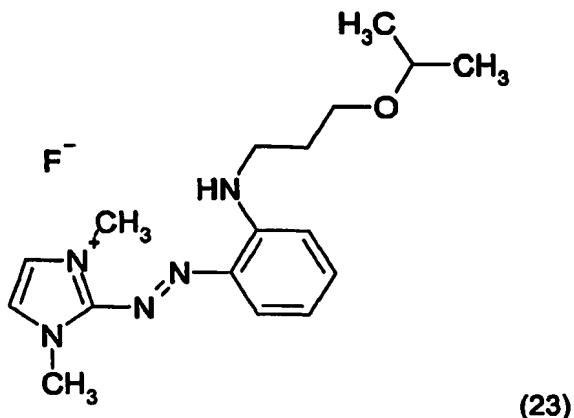
1g sodium methoxide is added under nitrogen atmosphere, at 293 K, to a stirred mixture of 10.0 ml of a 25% ammoniac solution, 100g isopropanol, to the compound of formula (21). Then, the temperature is raised to 333 K. The reaction mixture is stirred 28 hour at this temperature under the autoclave pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 45 ml of isopropanol and dried in vacuum to obtain 17.6 g of the compound of formula (20)

Example A2



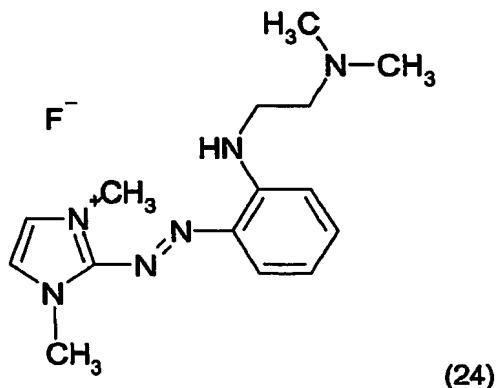
Under nitrogen atmosphere, at 293 K, to a stirred mixture of 10.0 ml of a 40% methylamine solution, 100g isopropanol, the compound of formula (21) is added. Then, the temperature is raised to 333 K. The reaction mixture is stirred 8 hour at this temperature under the autoclave pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 45 ml of isopropanol and dried in vacuum to obtain 18.6 g of product with the given formula (22).

Example A3



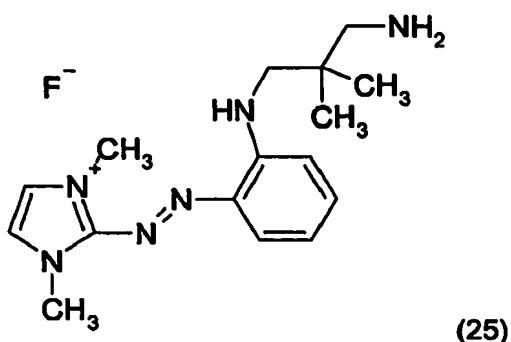
10 g triethylamine is added under nitrogen atmosphere, at 293 K, to a stirred mixture of 8 g 3-isopropoxy-1-propylamine, 100 g water and 27 g of compound of formula (21). Then, the temperature is raised to 333 K. The reaction mixture is stirred 3 hour at this temperature at normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 45 ml of acetonitrile and dried in vacuum to obtain 30,5 g of product with the given formula (23).

Example A4



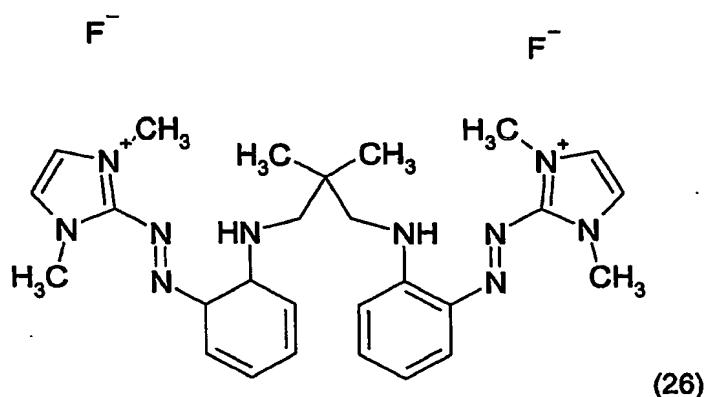
Under nitrogen atmosphere, at 293 K, to a stirred mixture of 1,8 g N,N-dimethylamino-ethylamin, 40 g acetonitrile, 6,6 g of compound of formula (21) is added. Then, the temperature is raised to 333 K. The reaction mixture is stirred 3 hour at this temperature under normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 15 ml of acetonitrile and dried in vacuum to obtain 9,0 g of product with formula (24).

Example A5



Under nitrogen atmosphere, at 293 K, to a stirred mixture of 3,06 g 2,2-dimethyl-1,3-diaminopropane and 40g water 11,3 g of compound of formula (21) is added. Then, the temperature is raised to 333 K. The reaction mixture is stirred 6 hour at this temperature under nitrogen and normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 30 ml of water/sodiumchloride solution and dried in vacuum to obtain 12 g of product with formula (25).

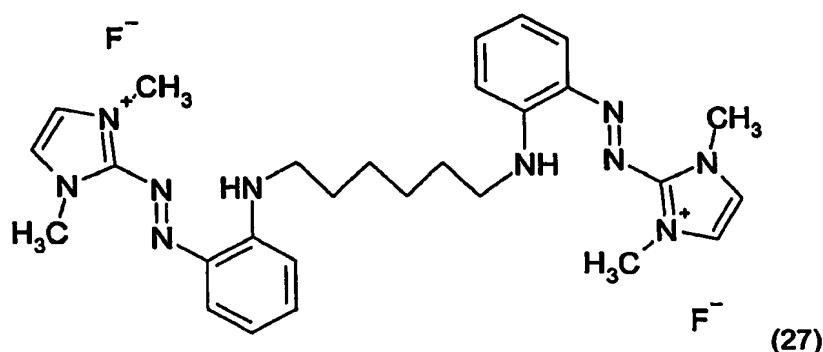
Example A6



4g triethylamine is added under nitrogen atmosphere, at 293 K, to a stirred mixture of 2,04 g 2,2-dimethyl-1,3-diamino-propane, 100g isopropanol, and 14,3 g of the compound of formula (21). Then, the temperature is raised to 333 K. The reaction mixture is stirred 6 hour at this temperature under nitrogen and normal pressure. Then, the reaction mass is stirred for 4

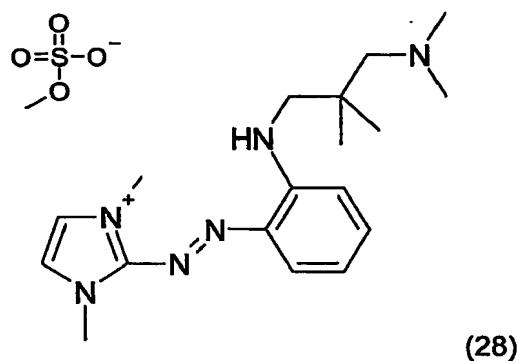
hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 30 ml of isopropanol and dried in vacuum to obtain 12 g of product of formula (26).

Example A7



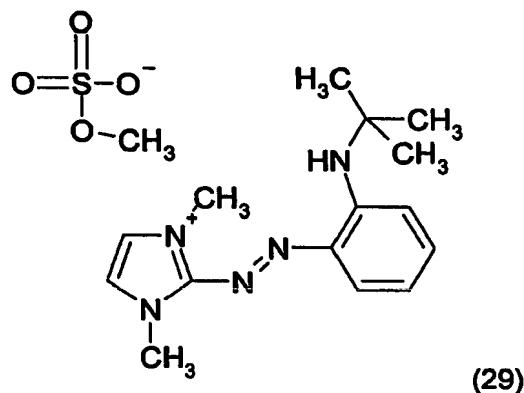
4g triethylamine is added under nitrogen atmosphere, at 293 K, to a stirred mixture of 2,3 g 1,6-diamino-hexane, 100g isopropanol, and 14,3 g of the compound of formula (21). Then, the temperature is raised to 343 K. The reaction mixture is stirred 16 hour at this temperature under nitrogen and normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 30 ml of isopropanol and dried in vacuum to obtain 10.6 g of product with the given formula (27).

Example A8



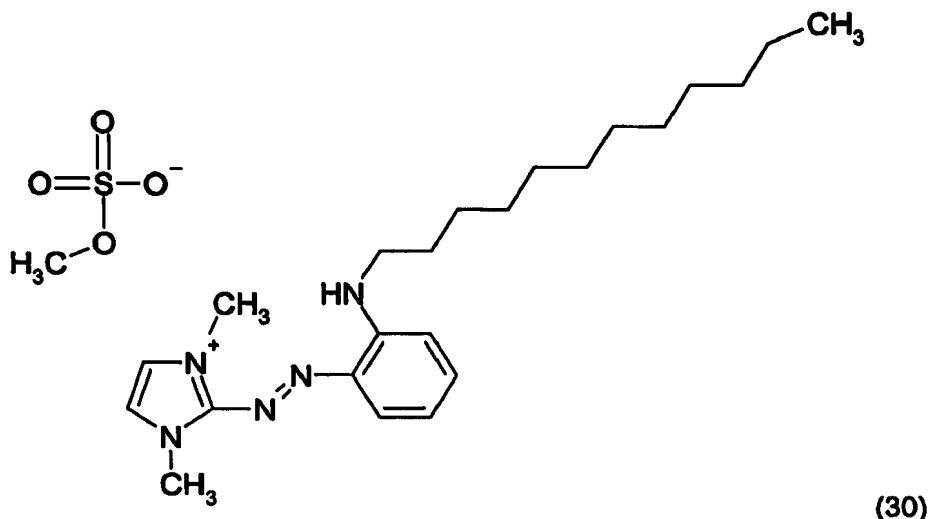
Under nitrogen atmosphere, at 293 K, add to a stirred mixture of 5,2 g 2,2,N,N-tetramethyl-1,3-diamino-propane, 100g acetonitrile, and 14,3 g of the compound of formula (21). Then, the temperature is raised to 333 K. The reaction mixture is stirred 3 hour at this temperature under nitrogen and normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 30 ml of acetonitrile and dried in vacuum to obtain 10.6 g of product of formula (28).

Example A9



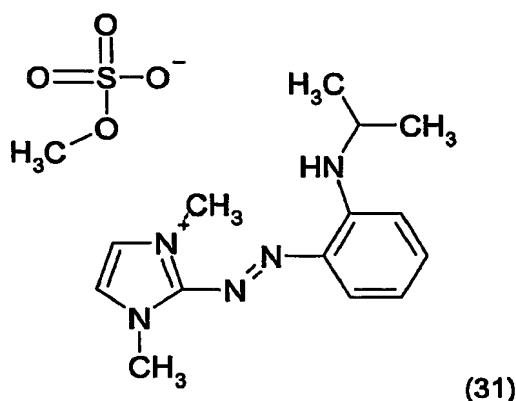
4g triethylamine is added under nitrogen atmosphere, at 293 K, to a stirred mixture of 2,3 g tert-butylamine, 40g acetonitrile, and 14,3 g of the educt of compound of formula (21). Then, the temperature is raised to 333 K. The reaction mixture is stirred 16 hour at this temperature under nitrogen and normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 15 ml of acetonitrile and dried in vacuum to obtain 10.6 g of product of formula (29).

Example A10



1g sodium methoxide is added under nitrogen atmosphere, at 293 K, to a stirred mixture of of 7,3 g dodecylamine, 100g acetonitrile, and 14,3 g of the compound of formula (21). Then, the temperature is raised to 333 K. The reaction mixture is stirred 8 hour at this temperature under normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 45 ml of acetonitrile and dried in vacuum to obtain 15.6 g of product with the given formula (30).

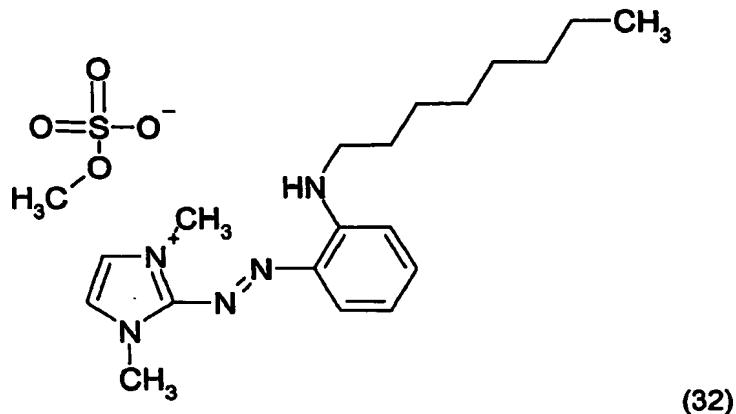
Example A11



Under nitrogen atmosphere, at 293 K, to a stirred mixture of 4,3 g isopropylamine, 40g water, and 14,3 g of the compound of formula (21) is added. The reaction mixture is stirred 8 hour

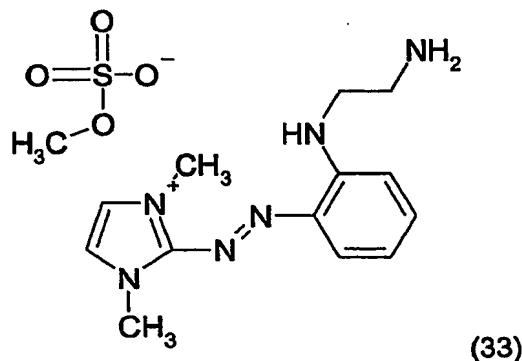
at this temperature under normal pressure. The reaction mass is filtered and the filter residue is washed with 45 ml of acetonitrile and dried in vacuum to obtain 10.6 g of product of formula (31).

Example A12



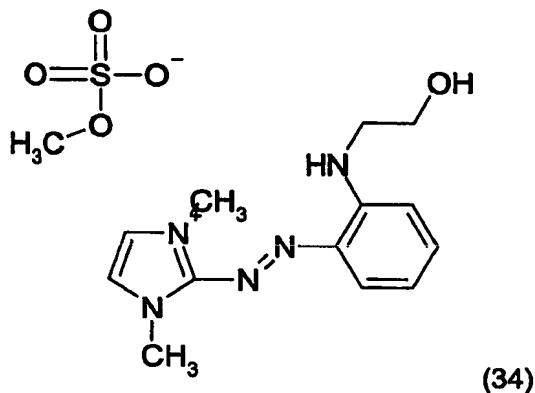
1g sodium methoxide is added under nitrogen atmosphere, at 293 K, to a stirred mixture of 6 g 1-octylamine, 100g isopropanol, to the 14 g of the compound of formula (21). Then, the temperature is raised to 333 K. The reaction mixture is stirred 8 hour at this temperature under normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 45 ml of isopropanol and dried in vacuum to obtain 14.8 g of product with the given formula (32).

Example A13



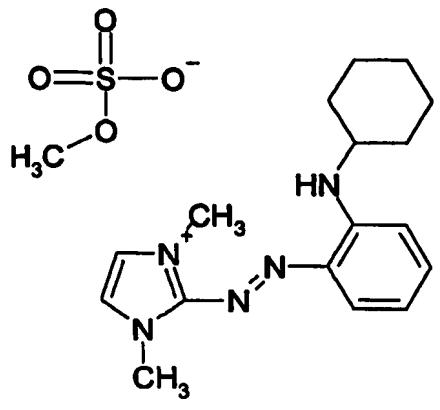
Under nitrogen atmosphere, at 293 K, to a stirred mixture of 3,8 g ethylenediamin, 40 g acetonitrile, 14 g of compound of formula (21) is added. Then, the temperature is raised to 333 K. The reaction mixture is stirred 3 hour at this temperature under normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 15 ml of acetonitrile and dried in vacuum to obtain 9,0 g of product with the given formula (33).

Example A14



Under nitrogen atmosphere, at 293 K, to a stirred mixture of 4 g triethylamine, 2,0 g ethanolamin, 40 g acetonitrile, 14 g of the compound of formula (21) is added. Then, the temperature is raised to 333 K. The reaction mixture is stirred 5 hour at this temperature under normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 15 ml of acetonitrile and dried in vacuum to obtain 9,0 g of product with the given formula (34).

Example A15



(35)

Under nitrogen atmosphere, at 293 K, to a stirred mixture of 4 g triethylamine, 3,8 g cyclohexylamin, 40 g acetonitrile, 14 g of the compound of formula (21) is added. Then, the temperature is raised to 333 K. The reaction mixture is stirred 8 hour at this temperature under normal pressure. Then, the reaction mass is stirred for 4 hours, while the temperature is decreased to 295 K. The reaction mass is filtered and the filter residue is washed with 15 ml of acetonitrile and dried in vacuum to obtain 9,0 g of product with the given formula (35).

Example B/ Application:

Example B1

A strongly alkaline 10 % solution of a non-ionic surfactant (Plantaren 2000, Henkel) is adjusted to pH 9.5 using citric acid. 0.1 % of the dye of formula (28) as given in example A8 is dissolved therein and a strand of human hair, bleached white, is treated with the dye solution at room temperature. After only a short period of time, the strand has been dyed a bluish-red shade, which is still very intensive even shampooing ten times. The dye also has a strong affinity to undamaged hair. In that case, too, the wash fastness is very good. The light fastness on damaged and undamaged hair is excellent. The perm fastness is on un- and damaged hair very good as well.

Example B2:

A 10 % solution of a non-ionic surfactant (Plantaren 2000, Henkel) is adjusted to pH 5.5 using citric acid. 0.1 % of the dye of formula (30) as given in example A10 is dissolved therein and a strand of middle blonde undamaged human hair is treated with the dye solution at room temperature. After only a short period of time, the strand has been dyed in a bluish shade, which has a good wash, perm and light fastness.

Example B3:

A dye emulsion, containing

0.1 % of the dye according to the invention of formula (26) as given in example A6, and

3.5 % cetearyl alcohol

1.0 % ceteareth 80

0.5 % glyceryl mono-di-stearate

3.0 % stearamide DEA

1.0 % stearamphopropyl sulfonate

0.5 % polyquaternium-6 and

water ad 100%

is applied for 30 minutes, at room temperature, to bleached human hair, and rinsed. The result is a very attractive vibrant red dyeing with good fastnesses.

Example B4

A dye emulsion, containing 1 % of the dye of formula (20) as given in example 1, pH= 9.8

Cetylstearylalcohol	11.0
	0
Oleth-5	5.0
Oleic acid	2.5
Stearic acid	2.5
monoethanolamide	
Coco fatty acid	2.5
monoethanolamide	
Sodium laurylsuphate	1.7
1,2-Propanediol	1.0
Ammoniumchloride	0.5
EDTA, Tetrasodiumsalt	0.2
Perfume	0.4
Comproteinhydrolysate	0.2
Silica	0.1

is mixed with the same weight of 6 % hydrogen peroxide solution and the mixture is immediately applied to a tress of brown hair. After 30 minutes the tress is rinsed, shampooed, rinsed and dried.

The color result is a very brilliant red shade.

Example B5:

A dye emulsion, containing 0.5 % of the dye of formula (20) as given in example 1, pH= 9.8

Cetylstearylalcohol	11.0
	0
Oleth-5	5.0
Oleic acid	2.5
Stearic acid	2.5
monoethanolamide	
Coco fatty acid	2.5
monoethanolamide	
Sodium laurylsuphate	1.7
Sodiumsulphite	1.0
Ascorbic acid	0.5
1,2-Propanediol	1.0
Ammoniumchloride	0.5
EDTA, Tetrasodiumsalt	0.2
Perfume	0.4
Cornproteinhydrolysate	0.2
Silica	0.1
Toluene-2,5-Diamine Sulfate	0.07
Resorcinol	0.02
2-Amino-6-Chloro-4-	0.01
Nitrophenol	
4-Amino-m-Cresol	0.03
2-Amino-3-Hydroxypyridine	0.00

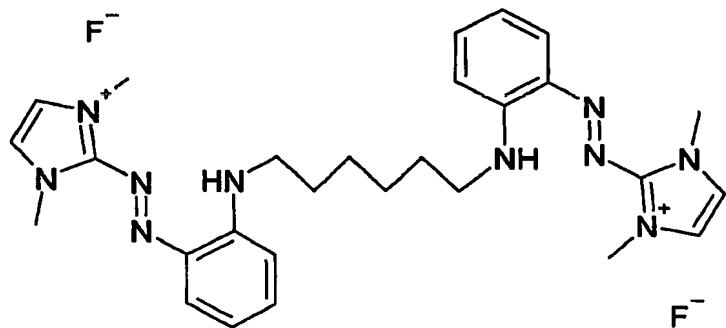
is mixed with the same weight of 6 % hydrogen peroxide solution and the mixture is immediately applied to a tress of brown hair. After 30 minutes the tress is rinsed, shampooed, rinsed and dried.

The color result is a very brilliant violet shade.

The advantage of the dyes of example 1, 2 and 3 is the stability against reduction agents e.g. sodium sulfite and ascorbic acid. Therefore you can combine them with oxidation dyes in one emulsion.

Example B6:

A strongly alkaline 10 % solution of a non-ionic surfactant (Plantaren 2000, Henkel) is adjusted to pH 9.5 using citric acid. 0.1 % of the dye of formula (20) as given in example 1 and 0.1 % of the dye formula



are dissolved therein and a strand of middle blonde undamaged human hair is treated with the dye solution at room temperature. After only 10 minutes, the strand has been dyed a intensive violett shade, which has a good wash, perm and light fastness.

Example B7:

A strongly alkaline 10 % solution of a non-ionic surfactant (Plantaren 2000, Henkel) is adjusted to pH 9.5 using citric acid. 0.1 % of the dye of example 1 and 0.1 % of the dye formula (25) as given in example A5 are dissolved therein and a strand of dark blonde undamaged human hair is treated with the dye solution at room temperature. After 20 minutes, the strand has been dyed a intensive aubergine shade, which has a good wash, perm and light fastness.

Example B8:**First step:**

A strand bleached blond hair is treated with a composition comprising 5 g of 6% by weight hydrogenperoxide solution and 5 g a composition comprising an oxidation base with a pH value of 9.8 as given below:

Composition comprising an oxidation base with a pH value of 9.8

10.0 % by weight olic acid	
toluene-2,5-diamine sulfate	0.07
resorcinol	0.02
2-amino-6-chloro-4-nitrophenol	0.01
4-amino-m-cresol	0.03
2-amino-3-hydroxypyridine	0.001
sodium sulfite	1.0
ascorbic acid	0.5
water	ad 100

Example B8a/Second Step:

After 15 minutes, the pH of the hair is adjusted to pH 5 by addition of citric acid. Then, 5 g of a 12,5% citric acid gel, comprising the 0.1% by weight of dye of formula (20) of example 1 according to the present application, is applied on the hair and combed, so that the hair has a pH of 7. After 15 minutes the hair is washed with water, rinsed and dried. The strand has been dyed in an intensive shade, which has a good wash and light fastness.

Example B8a/Second Step:

After 15 minutes, without rinsing, blond hair is treated with a composition comprising 5 g of 6% by weight hydrogenperoxide solution and 5 g a composition comprising an oxidation base with a pH value of 5 as given below:

Composition comprising an oxidation base with a pH value of 5 (adjusted with citric acid)

10.0 % by weight olic acid	
toluene-2,5-diamine sulfate	0.07

resorcinol	0.02
2-amino-6-chloro-4-nitrophenol	0.01
4-amino-m-cresol	0.03
2-amino-3-hydroxypyridine	0.001
sodium sulfite	1.0
ascorbic acid	0.5
water	ad 100

and 5 g of a 12,5% citric acid gel, comprising the 0.1% by weight of dye of formula (23) of example A3 according to the present application,

The hair is combed, so that the hair has a pH of 7. After 15 minutes the hair is washed with water, rinsed and dried. The strand has been dyed in an intensive shade, which has a good wash and light fastness.

Example B8c/Second Step:

After 15 minutes the hair is treated with 5 g composition comprising an oxidation base with a pH value of 9.8 as given below:

Composition comprising an oxidation base with a pH value of 9.8

10.0 % by weight olic acid	
toluene-2,5-diamine sulfate	0.07
resorcinol	0.02
2-amino-6-chloro-4-nitrophenol	0.01
4-amino-m-cresol	0.03
2-amino-3-hydroxypyridine	0.001
sodium sulfite	1.0
ascorbic acid	0.5
water	ad 100

Example B8a/Second Step:

After 15 minutes, the pH of the hair is adjusted to pH 5 by addition of citric acid. Then, 5 g of a 12,5% citric acid gel, comprising the 0.1% by weight of dye of formula (20) of example 1

according to the present application, is applied on the hair and combed, so that the hair has a pH of 7. After 15 minutes the hair is washed with water, rinsed and dried. The strand has been dyed in an intensive shade, which has a good wash and light fastness.

Example B8a/Second Step:

After 15 minutes, without rinsing, blond hair is treated by a comb with a composition comprising 5 g of 6% by weight hydrogenperoxide solution and 5 g of a composition comprising an oxidation base with a pH value of 9.8 as given below:

Composition comprising an oxidation base with a pH value of 9.8

10.0 % by weight olic acid	
toluene-2,5-diamine sulfate	0.07
resorcinol	0.02
2-amino-6-chloro-4-nitrophenol	0.01
4-amino-m-cresol	0.03
2-amino-3-hydroxypyridine	0.001
sodium sulfite	1.0
ascorbic acid	0.5
water	ad 100

and a solution comprising 1% by weight of a dye of formula (23) according to example A3 of the present application with a pH of 9.8. After 15 minutes the hair is washed with water, rinsed and dried. The strand has been dyed in an intensive shade, which has a good wash and light fastness.